

## Biology

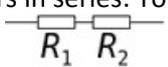
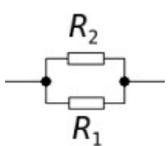
Microscopy	
Actual size =	$\frac{\text{image size}}{\text{magnification}}$

## Chemistry

General	
Number of neutrons = nucleon (mass) number – atomic number	
Quantitative chemistry - Moles	
SOLIDS: Number of moles ( <i>mol.</i> ) = $\frac{\text{mass (g)}}{A_r \text{ (g mol}^{-1}\text{)}}$ or $\frac{\text{mass (g)}}{M_r \text{ (g mol}^{-1}\text{)}}$	$n = \frac{m}{A_r} = \frac{m}{M_r}$
LIQUIDS: Number of moles ( <i>mol.</i> ) = concentration ( <i>mol dm<sup>-3</sup></i> ) × volume ( <i>dm<sup>3</sup></i> )	$n = CV$
LIQUIDS: Concentration ( <i>mol dm<sup>-3</sup></i> ) × volume ( <i>dm<sup>3</sup></i> ) = concentration ( <i>mol dm<sup>-3</sup></i> ) × volume ( <i>dm<sup>3</sup></i> )	$C_1V_1 = C_2V_2$
GASES: Volume of a gas ( <i>m<sup>3</sup></i> ) = number of moles ( <i>mol.</i> ) × 24 ( <i>m<sup>3</sup> mol<sup>-1</sup></i> ) (at room temperature and pressure)	$V = 24n$

## Physics

General	
Average speed ( <i>ms<sup>-1</sup></i> ) = $\frac{\text{distance (m)}}{\text{time (s)}}$	
Average velocity ( <i>ms<sup>-1</sup></i> ) = $\frac{\text{displacement (m)}}{\text{time (s)}}$	$v = \frac{s}{t}$
Acceleration ( <i>ms<sup>-2</sup></i> ) = $\frac{\text{final velocity (ms}^{-1}\text{)} - \text{initial velocity (ms}^{-1}\text{)}}{\text{time (s)}}$	$a = \frac{v-u}{t}$
Weight ( <i>N</i> ) = mass ( <i>kg</i> ) × gravitational field strength ( <i>ms<sup>-2</sup></i> ) <i>Note: Earth's gravitational field strength = 10 ms<sup>-2</sup></i>	$F = mg$
Force ( <i>N</i> ) = mass ( <i>kg</i> ) × acceleration ( <i>ms<sup>-2</sup></i> )	$F = ma$
Density ( <i>kg m<sup>-3</sup></i> ) = $\frac{\text{mass (kg)}}{\text{volume (m}^3\text{)}}$	$\rho = \frac{M}{V}$
Hooke's law: Force ( <i>N</i> ) = constant ( <i>N m<sup>-1</sup></i> ) × extension ( <i>m</i> )	$F = kx$
Pressure ( <i>Pa</i> ) = $\frac{\text{force (N)}}{\text{area (m}^2\text{)}}$	$P = \frac{F}{A}$
Fluid Pressure ( <i>Pa</i> ) = density ( <i>kg m<sup>-3</sup></i> ) × gravitational field strength ( <i>ms<sup>-2</sup></i> or <i>N kg<sup>-1</sup></i> ) × height ( <i>m</i> )	$P = \rho gh$
Work ( <i>J</i> ) = force ( <i>N</i> ) × distance moved ( <i>m</i> )	$\Delta E = Fd$
Power ( <i>W</i> ) = $\frac{\text{work (J)}}{\text{time (s)}}$	$P = \frac{\Delta E}{t}$
Kinetic Energy ( <i>J</i> ) = $\frac{1}{2} \times \text{mass (kg)} \times \text{velocity}^2 \text{ (ms}^{-1}\text{)}$	$KE = \frac{1}{2}mv^2$
Gravitational potential energy ( <i>J</i> ) = mass ( <i>kg</i> ) × gravitational field strength ( <i>ms<sup>-2</sup></i> or <i>N kg<sup>-1</sup></i> ) × height ( <i>m</i> )	$GPE = mgh$
Efficiency (%) = $\frac{\text{useful power output (W)}}{\text{total power input (W)}} \times 100$	$\text{Efficiency} = \frac{P_{\text{out}}}{P_{\text{in}}}$
Efficiency (%) = $\frac{\text{useful energy output (J)}}{\text{total energy input (J)}} \times 100$	$\text{Efficiency} = \frac{E_{\text{out}}}{E_{\text{in}}}$
Moment ( <i>Nm</i> ) = force ( <i>N</i> ) × perpendicular distance from pivot ( <i>m</i> )	$M = Fd$
Sum of clockwise moments ( <i>Nm</i> ) = sum of anticlockwise moments ( <i>Nm</i> )	$F_1d_1 = F_2d_2$

Thermal	
Boyle's Law for changes in gas pressure at constant temperature : $\text{pressure}_1 (\text{Pa}) \times \text{volume}_1 (\text{m}^3) = \text{pressure}_2 (\text{Pa}) \times \text{volume}_2 (\text{m}^3)$ <p style="text-align: center;">or</p> $\text{pressure} (\text{Pa}) \times \text{volume} (\text{m}^3) = \text{constant}$	$P_1 V_1 = P_2 V_2$ <p style="text-align: center;">or</p> $PV = \text{constant}$
Energy (J) = mass (kg) × specific heat capacity (Jkg <sup>-1</sup> °C <sup>-1</sup> ) × temperature change (°C)	E = mcΔT
Electricity	
Current (A) = $\frac{\text{charge (C)}}{\text{time (s)}}$	$I = \frac{Q}{t}$
Voltage (V) = $\frac{\text{energy transferred (J)}}{\text{charge (C)}}$	$V = \frac{E}{Q}$
Voltage (V) = current (A) × resistance (Ω)	V = IR
Power (W) = current (A) × voltage (V)	P = IV
Power (W) = current <sup>2</sup> (A) × resistance (Ω)	P = I <sup>2</sup> R
Energy transferred (J) = current (A) × voltage (V) × time (s)	ΔE = IVt
Energy transferred (J) = power (W) × time (s)	ΔE = Pt
Resistors in series: Total Resistance (Ω) = sum of individual resistors (Ω) 	R <sub>TOTAL</sub> = R <sub>1</sub> + R <sub>2</sub> + R <sub>3</sub> + ... R <sub>n</sub>
Resistors in parallel: $\frac{1}{\text{total resistance (Ω)}} = \frac{1}{\text{sum of individual resistors (Ω)}}$ 	$\frac{1}{R_{TOTAL}} = \frac{1}{R_1} + \frac{1}{R_2} \dots \frac{1}{R_n}$
Resistance (Ω) = $\frac{\text{resistivity (Ωm)} \times \text{length (m)}}{\text{area (m}^2\text{)}}$ <i>Note: since wires have a circular cross section, area = π × radius<sup>2</sup></i>	$R = \frac{\rho l}{A}$
Transformers: $\frac{\text{voltage in secondary coil (V)}}{\text{voltage in primary coil (V)}} = \frac{\text{turns on secondary coil}}{\text{turns on primary coil}}$	$\frac{V_s}{V_p} = \frac{N_s}{N_p}$
Transformers: $\frac{\text{voltage in primary coil (V)}}{\text{voltage in secondary coil (V)}} = \frac{\text{current in secondary coil (A)}}{\text{current in primary coil (A)}}$	$\frac{V_p}{V_s} = \frac{I_s}{I_p}$
Waves	
Wave speed (ms <sup>-1</sup> ) = frequency (Hz) × wavelength (m)	c = fλ
Frequency (Hz) = $\frac{1}{\text{Period (s)}}$	$f = \frac{1}{T}$
Refractive index = $\frac{\text{sine of the angle of incidence, } i}{\text{sine of the angle of refraction, } r}$	$n = \frac{\sin i}{\sin r}$
Refractive index = $\frac{\text{speed of light in vacuum}}{\text{speed of light in material}}$	$n = \frac{c_v}{c_m}$
Refractive index = $\frac{1}{\text{sine of critical angle}}$	$n = \frac{1}{\text{sinc}}$
Nuclear	
Radioactive alpha decay: ${}_{92}^{238}\text{Th} \rightarrow {}_{90}^{234}\text{U} + {}_2^4\text{He} + \text{energy}$	${}_Z^A X \rightarrow {}_{Z-2}^{A-4} Y + {}_2^4\text{He}$
Radioactive beta decay: ${}_{82}^{209}\text{Pb} \rightarrow {}_{83}^{209}\text{Bi} + {}_{-1}^0\text{e} + \text{energy}$	${}_Z^A X \rightarrow {}_{Z+1}^A Y + {}_{-1}^0\text{e}$
Radioactive gamma decay: ${}_{27}^{60}\text{Co} \rightarrow {}_{27}^{60}\text{Co} + \gamma + \text{energy}$	${}_Z^A X \rightarrow {}_Z^A Y + \gamma$
Energy (J) = mass defect (kg) × speed of light <sup>2</sup> (ms <sup>-1</sup> )	E = mc <sup>2</sup>

